The University of Zambia School of Engineering Dept. of Civil & Environmental Engineering

CEE 4412: Environmental Engineering I

WASTEWATER/FAECAL SLUDGE MANAGEMENT

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Objectives

- The aim of this topic is to introduce students to the aspects of wastewater management. Specific objectives include to:
 - Introduce students to wastewater and its characteristics
 - Highlight environmental and public health implications of Wastewater
 - Explain some of the amelioration measures for addressing the environmental and public health impacts associated with wastewater

Sanitation

Before we delve into wastewater management, we firstly

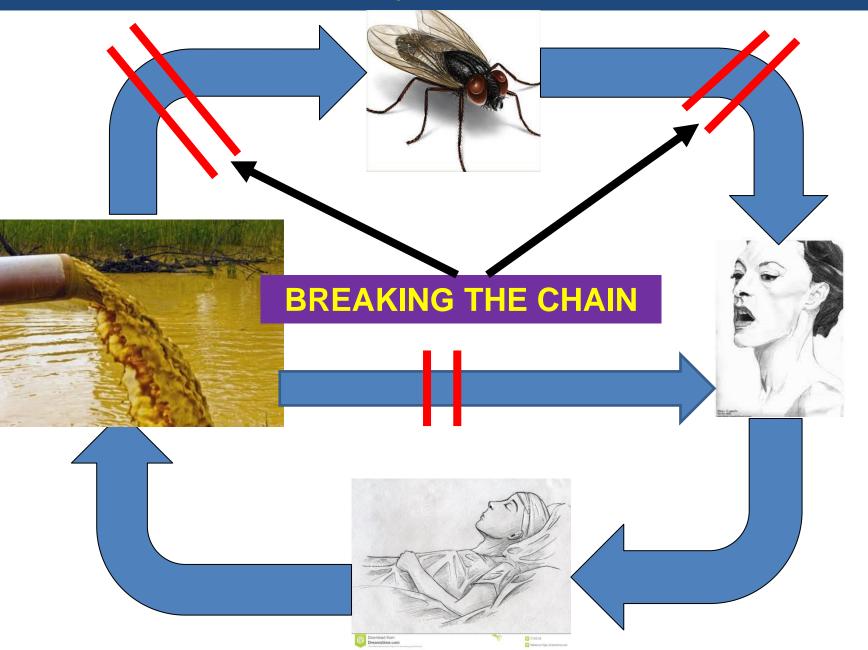
look at sanitation which means:

Making Healthy

or

Removal and safe disposal of wastes that can be hazardous to health

Sanitation – Why??



- 1. Wastewater Management
- 2. Faecal sludge or Excreta management (faeces, urine)
- 3. Greywater management
- 4. Solid waste management
- 5. Drainage (for rainwater / stormwater)

Solid waste management



Drainage for storm water







What is wastewater?

Water that is no longer needed – mostly which has served the intended purpose and in the process has been loaded with various pollutants in form of suspended, colloidal and dissolved constituents.

Or

Any water that has been adversely affected in quality by anthropogenic activities and needs to be disposed of.



Demestic sewage or pure sewage (Excreta Sullage)
 Trade efficients
 Faeces + Urine (Black Water)
 Industrial

Sewage + Trade = Municipal Wastewater

Cleaning of food; Food left-overs
 Dish washing; Body washing; Washing of clothes
 Washing of floors (Grey Water)

Characteristics

Organic Matter (CHNOP)

- High content of organic matter mainly in form of:-
 - Carbohydrates
 - Fats and Grease
 - Protein

Raw Sewage Strength (BOD)

- Characterisation of wastewater is usually through the
- organic matter content as follows:

| WEAK | MEDIUM | STRONG |
|------|--------------|--------|
| >200 | <200 to >350 | >350 |

Determination of strength is either through Chemical Oxygen Demand or Biochemical Oxygen Demand.

Measurement of Organic Matter (Strength)

Chemical Oxygen Demand (COD)

 $H_2SO_4+SAMPLE+K_2Cr_2O_7 \rightarrow H_2O+CO_2+NO_3$

This test will give the oxygen required to oxidise both the organic and non organic component of the wastewater

Biochemical Oxygen Demand (BOD)

Simulating Natural process

Organic matter +DO + Saprophytic bacteria = More
Bacteria +H₂O+CO₂ + Energy

• For domestic wastewater COD/BOD = 1.5 - 1.8

COD vs BOD importance of relationship

Suspended Solids

Wasterwater usually has a lot of SS most of which form part of the COD-BOD.

COD-BOD can be in form of dissolved, colloidal or suspended matter

NOTE: Suspended matter that settles in 1 hour
 =Settleable solids

Dissolved Gases

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- DO Low to absent (measure of degree of treatment)
 - NH₃ high due to biodegradation of protein
 - CO₂ high due to biodegradation process of organic matter
 - H_2S High due to anaerobic decomposition of Sulphur containing substances
 - CH_4 where anaerobic digestion has been allowed to go on for some time (Bio-gas)

Microbiological characteristics

Contains all microorganisms being excreted by the population



Industrial Wastewater

 Differs from domestic as each is different (e.g. acidic from mines and alkaline from textile and tannery).
 Significance: Can inhibit growth of microorganisms required in the treatment process



Types of industrial wastewater

- Cooling water (biggest volume; least polluted)
- Rinse water (from food packaging companies not so polluted)
- Process water (small volumes with high concentration of pollutants)

Process Water

- High organic content
- Extreme pH
- toxic

Process Water - Types

With high organic content

≻High COD; High BOD e.g. breweries

≻High COD; Low BOD

With extreme pH

- e.g. acidic from mines and alkaline from textile and tannery.
- Significance: Can inhibit growth of microorganisms required in the treatment process

- Like heavy metals and chemicals which will be:-
- Harzadous to human health
- Dangerous to fish
- Impair functioning of microorganisms in natural water bodies and treatment plants

Reasons for wastewater treatment

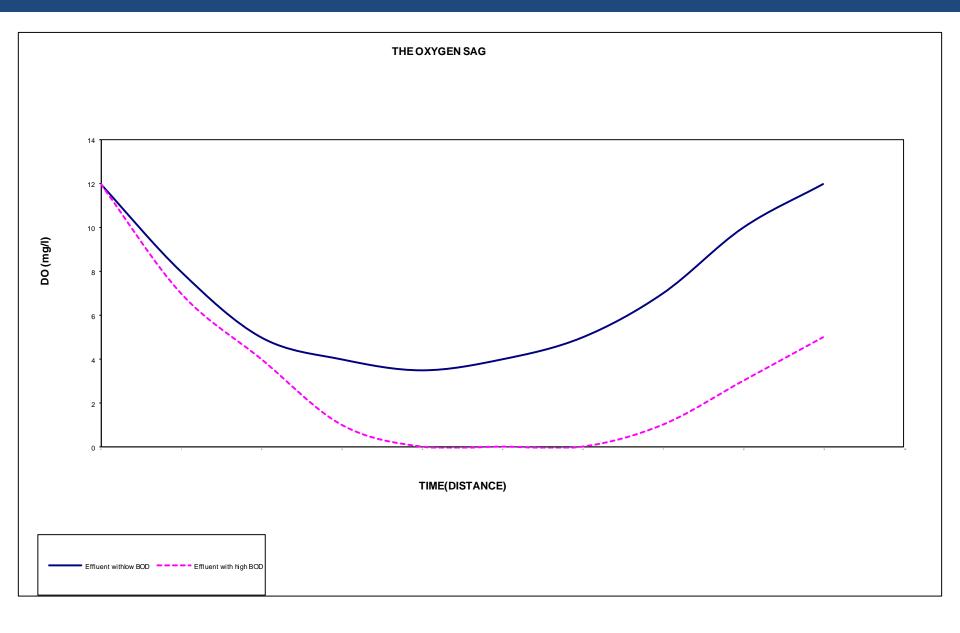


Prevent Depletion of O₂

- ✤ Aquatic creatures (fish etc) need DO
- Untreated wastewater discharged into a river introduces organic matter
- Bacteria metabolise organic matter using DO
- Bacteria increase in no. using up more DO
- Depletion of O₂ occurs
- Death of the aquatic lif



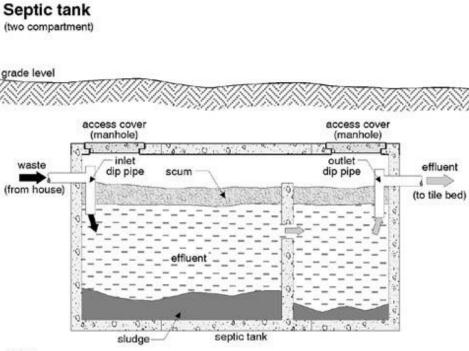
Oxygen sag versus BOD concentration



Reasons for wastewater treatment cont'

> REDUCING SLUDGE AND

- Avoid formation of sludge banks
- Avoid sludge deposits at the bottom of the receiving water body
- Avoid formation of scum in the receiving water body
- Avoid unsightliness and odour problems and
- Avoid oxygen depletion that may arise from scum and sludge.



Prevent III-effects on Human Health

- Wastewater contains disease causing bacteria (pathogens), viruses and worm eggs.
- Many serious outbreaks of diseases e.g. typhoid/cholera can be traced to the contamination of drinking water by wastewater/excreta
- Treating wastewater helps prevent the spread of diseases

Prevent nutrient loading to the river

Increased growth of algae in river water due to nutrients in the water might occur (eutrophication).





E.G effects of ww on veg.-Mindolo kitwe (2012)



WW reuse in agriculture and horticulture

Aesthetic Reasons

The clarity and colour of the water may be affected

Taste and odour problems

Dead?

Cost Implications

Untreated wastewater will introduce pathogens into receiving water bodies

- Additional treatment especially wrt chlorination at water treatment plant will be required (Chongwe, Kafubu, Pre-Chlorination, activated carbon)
- In communities drawing directly from river, this can lead to outbreaks (Medicines and morbidity/Mortality)
- Resulting nutrient loading will lead to algal blooms leading to increased treatment requirements at water treatment plant (Organic, clogging, odour)

Example- Chongwe Plant (2011)



Effluent Standards

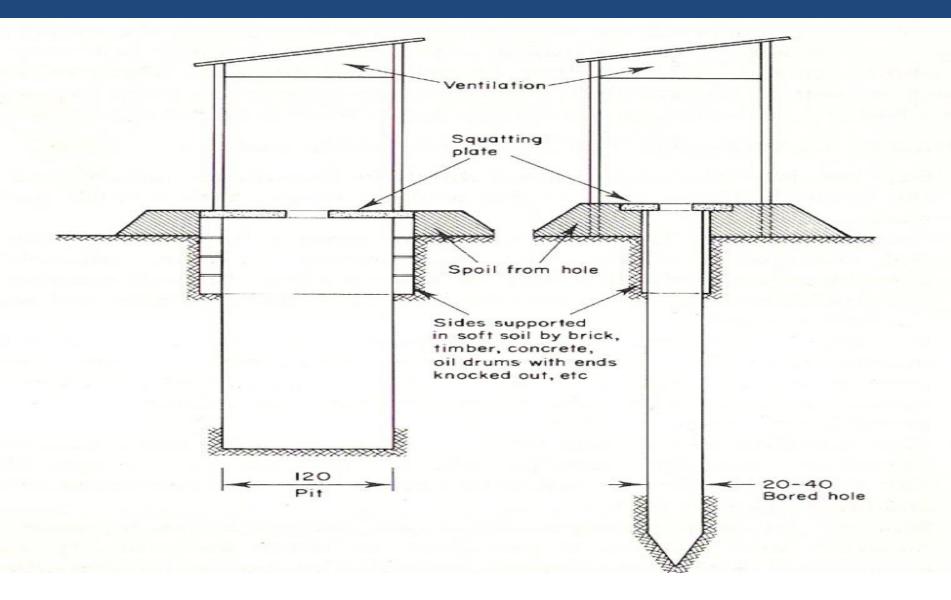
- ✤ BOD
- ✤ COD
- Settleable Solids in 2hrs
- ✤DO
- Temp
- Nitrate
- Total Phosphates

- =<50mg/l
- =< 90mg/l
- =< 0.5mg/l
- => 5mg/l
- =< 40deg cel
- =< 50mg/l
- =< 6mg/l

Types of Sanitation Systems

- ✤On-site
- ✤ Off-site

On-site/Drop-and-Store



Off-site/Flush-and-Forget

